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Note

Paper chromatography of ^{113m}In -labelled radiopharmaceuticals: [^{113m}In]DTPA

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The rapid growth of the use of short-lived radiopharmaceuticals in nuclear medicine is associated with the development of analytical techniques, such as chromatography, that can be used to identify various components in these products. Radiopharmaceuticals incorporating ^{113m}In ($T_{\frac{1}{2}} = 100$ min) are being widely used in organ scanning. The paper chromatography of ^{113m}In -labelled radiopharmaceuticals for brain, placenta, liver and spleen scanning is described in this paper.

METHODS AND MATERIALS

A 25 mCi ^{113m}In generator obtained from the Radiochemical Centre, Amersham, Great Britain, was used for the preparation of ^{113m}In -labelled radiopharmaceuticals. ^{113m}In was eluted as indium chloride with 6 ml of 0.04 *N* HCl. The following preparations were made according to published methods:

(i) ^{113m}In -gelatin complex for placental scanning¹.

(ii) [^{113m}In] indium hydroxide colloid for liver and spleen scanning².

(iii) [^{113m}In]DTPA chelate for brain scanning³. This was also prepared from a brain scanning kit (Code No. 80) supplied with the indium generator. The preparation was carried out by injecting 5 ml of generator eluate into a vial containing component A (a sterile aqueous solution of DTPA and acetic acid) followed by the injection of 1.0 ml of component B (a sterile aqueous buffer containing 31.7 mg of trometamol).

The preparations were chromatographed on Whatman No. 1 paper using the ascending technique in the following solvents: 0.1 *N* HCl; 0.1 *N* NH_4OH ; 0.1 *N* NH_4OAc ; 3% NaCl; and 85% methanol. The chromatograms were cut into 1 × 2 cm strips and counted in a Nuclear-Chicago automatic well counting system, Model 4219 (Nuclear-Chicago, Des Plaines, Ill., U.S.A.).

RESULTS AND DISCUSSION

For the chromatography of short-lived radiopharmaceuticals, it is desirable to use solvent systems that have rapid flow-rates. In general, mixtures of organic solvents provide good separations but have the disadvantage of slow runs. Aqueous solvents, which are usually fast, appear to be suitable for ^{113m}In -labelled radiopharmaceuticals.

The R_F values of various ^{113m}In -labelled preparations in different solvents are shown in Table I.

TABLE I
 R_F VALUES OF ^{113m}In -LABELLED RADIOPHARMACEUTICALS

Solvent	Time (h)	Solvent front (cm)	R_F^*			
			Indium chloride	Indium-gelatin complex	Indium hydroxide colloid	Indium-DTPA**
0.1 N HCl	1	19-21	0.92	0.90	0.75 (T)	0.93
0.1 N NH_4OH	1	19-20	0.03	0.03 (T)	0.03	0.31 (T)
0.1 N NH_4OAc	1	19-20	0.03	0.03	0.03	0.95
3% NaCl	1	18-19	0.03 (T)	0.03	0.03	0.95
85% Methanol	1.5	16-17	0.06	0.03	0.03	0.44

* T = tailing.

** Prepared according to Hill *et al.*³.

It can be seen from Table I that the R_F values for indium chloride, the indium-gelatin complex and indium hydroxide colloid preparations are very similar. The [^{113m}In] indium hydroxide colloid also shows an unidentified peak (<3%) along the solvent front in 3% NaCl.

[^{113m}In]DTPA is well separated in 0.1 N NH_4OAc , 3% NaCl and 85% methanol. [^{113m}In]DTPA as prepared from the brain scanning kit and by the procedure of Hill *et al.*³ showed similar chromatographic behaviour. A few preparations showed another small peak (<4%) in 85% methanol, which probably corresponds to indium hydroxide³. It can be concluded that the above solvent systems are suitable for determining the radiochemical purity of [^{113m}In]DTPA preparations.

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